

KIBABII UNIVERSITY

UNIVERSITY EXAMINATIONS 2022/2023 ACADEMIC YEAR

FOUR YEAR FIRST SEMESTER

FOR THE DEGREE OF BED (SCIENCE)

COURSE CODE: SCH 413

COURSE TITLE: POLYMER CHEMISTRY

DURATION: 2 HOURS

DATE: 17/04/2023 TIME: 9:00-11:00AM

INSTRUCTIONS TO CANDIDATES

Answer **QUESTION ONE** (Compulsory) and any other two (2) Questions.

- Indicate answered questions on the front cover.

Start every question on a new page and make sure question's number is written on each page.

This paper consists of 3 printed pages. Please Turn Over



QUESTION ONE (30 Marks)

1). a). Explain the process of vulcanization in polymerization.

(2marks)

b). Write the equations for the synthesis of the following polymer

(4marks)

- (i) Teflon
- (ii) PVC
- c) What is the role of potassium hydroxide in preparation of Nylon 6,10 by interfacial polymerization? (2marks)
- c). Distinguish between the following terms as used in polymer chemistry
- i. Polymer and monomer

(2marks)

ii). Functionality of polymer and Polydispersity index

(3marks)

iii). State two types of polymerization

(2marks)

d) State any four methods used in determining the molecular weight of a polymer

(4marks)

e). Giving examples state five types natural polymer

(5marks)

f) What is meant by degree of polymerization

(4marks)

g) Explain what is meant by glass transition temperature.

(2marks)

QUESTION TWO (20 Marks)

a) Name the following polymers according to the common nomenclature and then identify their monomers in respect to their polymers (6marks)

i.
$$(-CH_2-CH_2-)_n$$

 $-[-NH-(CH_2)_6-NHC-(CH_2)_8-C]_n$
ii. $(-CH_2-CH_2-)_n$

iii.

2

- b). State three polymer properties and characterization as used in polymer chemistry. (3marks)
- c) i. State two types of polythene

(2marks)

ii. Giving examples in each case, explain the difference between addition and condensation polymerization. (9 marks)

QUESTION THREE (20 Marks)

a). Explain the monomer arrangement in the co-polymer.

(10marks)

b). While giving the properties and application, explain how Nylon 6,6 is produced through condensation polymerization. (10 marks)

QUESTION FOUR (20Marks)

- a). Giving examples explain what is meant by polymer tacticity
- b) Write down the expressions for Mn and Mm and explain the conditions under which Mn = M_m. (10 marks)
- c). A polymer consists of 40% by mass of macromolecules of molecular mass 10000 and 60% by weight of macromolecules of molecular mass 75000. Calculate the number-average and mass-average polymer. molecular of the mass (10marks)

QUESTION FIVE (20Marks)

- a). While giving the advantages, disadvantages and application, Explain Bulk polymerization. (6marks)
- b) Explain the free radical mechanism for the polymerization of ethene.

(2 marks)

- c) List four ways of describing a polymer
- d). Differentiate thermoplastics and thermosetting plastics (6marks)

(5marks)

(6marks)