



KIBABII UNIVERSITY

UNIVERSITY EXAMINATIONS 2020/2021 ACADEMIC YEAR

FOURTH YEAR SECOND SEMESTER MAIN EXAMINATIONS

FOR THE DEGREE OF BACHELOR OF SCIENCE

COURSE CODE:

SCH 431

COURSE TITLE:

CHEMISTRY OF NATURAL PRODUCTS

DATE: 12/10/2021

TIME: 2:00-4:00PM

INSTRUCTIONS TO CANDIDATES

TIME: 2 Hours

Answer question ONE and any TWO of the remaining

KIBU observes ZERO tolerance to examination cheating

Question 1 (30 marks)

a) Define the following terminologies

[7 marks]

- i). Phytoallexin
- ii). Allelopathy
- iii). Myctoxin
- iv). Fatty acid
- v). Patenting
- vi). Bioprospecting
- vii). Aposematism
- b) The conversion of Acetyl-S-CoA to Malonyl-S-CoA is shown in the reaction given below

i). Give the structure of biotin

[2 marks]

ii). Biotin acts as a coenzyme. What is a coenzyme?

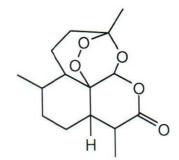
[1 mark]

- iii). Give the mechanism for the transformation of acetyl –S-CoA to malonyl–S-CoA using biotin as a coenzyme. [4 marks]
- iv). Starting form Acetyl-S-CoA, show how Butyl-S-ACP is formed. Use equations to illustrate this transformation. [7 marks]
- c) Mention four physiological activities of prostaglandins

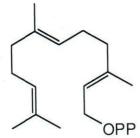
[4 marks]

Question 2 (20 marks)

a) Artemisinin is a drug used to treat multi-drug resistant strains of *Plasmodium falciparum* malaria. The compound is a sequiterpene lactone isolated from the plant *Artemisia* annua, long use in traditional Chinese medicine. Outline the biosynthesis of artemisinin starting from farnesyl diphosphate. [20 marks]



Artemisinin



Farnesyl diphosphate

Question 3 (20 marks)

a) Conversion of lanosterol to cholesterol involves loss of three methyl groups, saturation of the side chain and double bond migration of the $\mathfrak{L}^{8,9}$ double bond to the $\mathfrak{L}^{5,6}$ position. Using equations, show how the methyl groups in position 4 in lanosterol are lost, and how the side chain double bond is saturated. [8 marks]

b) Amino acids are divided into families depending on a common biosynthetic precursor. The aromatic amino acids, tryptophan, tyrosine and phenylalanine are derived from chorismic acid (structure shown below).

i). Show how the amino acid tyrosine and phenylalanine are formed from chorismic acid [6 marks]

$$HO$$
 NH_2
 NH_2
 NH_2
 NH_2
 NH_2
 NH_2

ii). Papaverine (structure shown below) found in opium poppy is an opium alkaloid antipasmodic drug, used primarily in the treatment of visceral spasm and vasospasm.

Starting from tyrosine, show how this alkaloid can be biosynthesized. [6 marks]

Question 4 (20 marks)

- a) Pyridoxal phosphate is a cofactor important in amino acid metabolism.
 - i). Show how the phosphate transforms the amino acid lysine to 4-aminobutanal

[8 marks]

ii). Show how aminobutanal can be transformed to lupinine and spartein (structures shown below) [6 marks]

$$CH_2OH$$

$$N$$

$$Lupinine$$
Spartein

b) E-Cinnamic acid is transformed to sinapic acid. Show how this can be achieved [6 marks]

Question 5 (20 marks)

 a) Most phospholipids contain a diglyceride, a phosphate group and a simple organic molecule. Starting from glycerol-3-phosphate, give a general scheme for the formation of phosphatidic acid. [4 marks]

Glycerol-3-phosphate

- b) Phosphatidic acid undergoes hydrolysis to give 1,2-diacylglycerol to which is then transformed to triglyceride. Using equations, show how this transformation is accomplished [5 marks]
- c) The phosphate group in phosphatidic acid is esterified with an alcohol to give phospholipids, such as phosphatidylcholine, phosphatidylethanolamine, phosphatidylserine and phosphatidylmyoinositiol. Give structures of these phospholipids. [4 marks]
- d) Give one source and one use of phosphatidylcholine.

[2 marks]

- e) Platelet Activating Factor (PAF) activates blood platelets and contributes to diverse biological effects
 - i). Give the structure of PAF

[2 marks]

ii). Identify the different biological effects PAF contributes to.

[3 marks]