



B

KIBABII UNIVERSITY

UNIVERSITY EXAMINATIONS
2020/2021 ACADEMIC YEAR

FIRST YEAR FIRST SEMESTER
MAIN EXAMINATIONS

FOR THE DEGREE OFBED (SCIENCE)

COURSE CODE: SPH115

COURSE TITLE: HEAT & PROPERTIES OF MATTER

DURATION: 2 HOURS

DATE: 17/02/2021

TIME: 2:00-4:00 PM

INSTRUCTIONS TO CANDIDATES

- Answer **QUESTION ONE** (Compulsory) and any other two (2) Questions.
- Indicate **answered questions** on the front cover.

Start every question on a new page and make sure question's number is written on each page
This paper consists of 3 printed pages. Please Turn Over

KIBU observes ZERO tolerance to examination cheating

SPH115: HEAT & PROPERTIES OF MATTER

QUESTION ONE [30 Marks]

- a) Distinguish between the concepts of heat and temperature. [2 Marks]
- b) State the zeroth law of temperature [1 Mark]
- c) State two parameters required for the establishment of a temperature scale [2 Marks]
- d) Explain three factors affecting changes in fixed points [3 Marks]
- e) Differentiate between steady flow and turbulent flow [2 Marks]
- f) A platinum resistance thermometer has a resistance of 10.40 ohms at 0°C and 14.35 ohms at 100°C . Assuming that the resistance changes uniformly with temperature, what is
- i) The temperature when the resistance is 11.19 ohms?
- ii) The resistance of the thermometer when the temperature is 45°C ? [4 Marks]
- g) What is the body temperature of a normal human being on the Celsius scale? What will this value be on the Fahrenheit scale? [2 Marks]
- h) Calculate the pressure exerted on the ground by a truck of mass 1600 kg if each wheel has an area of 0.02 m^2 in contact with the ground. [3 Marks]
- i) State two properties of an ideal liquid [2 Marks]
- j) Explain three factors that affect the surface tension of a liquid [3 Marks]
- k) Differentiate between density and viscosity [2 Marks]
- l) Explain why using a fan in the summer feels refreshing. [2 Marks]
- m) A student comes to school by a bicycle whose tire is filled with air at a pressure 240 kPa at 27°C . She travels 8 km to reach the school and the temperature of the bicycle tire increases to 39°C . What is the change in pressure in the tire when the student reaches school? [3 Marks]

QUESTION TWO [20 Marks]

- a) What is the total rate of radiation of energy from a human body with surface area 1.2 m^2 and surface temperature $30^{\circ}\text{C}=303\text{ K}$ if the surroundings are at a temperature of 20°C , what is the net rate of radiative heat loss from the body? The emissivity of the human body is very close to unity, irrespective of skin pigmentation [6 Marks]
- b) A polystyrene foam icebox has a total area of 0.950 m^2 and walls with an average thickness of 2.50 cm. The box contains ice, water, and canned beverages at 0°C . The inside of the box is kept cold by melting ice. How much ice melts in one day if the icebox is kept in the trunk of a car at 35.0°C ? [Take $k=0.010\text{ W/m}\cdot^{\circ}\text{C}$] [8 Marks]
- c) Briefly describe the three mechanisms of heat transfer giving an example in each case [6 Marks]

QUESTION THREE [20 Marks]

- a) Define the following terms giving their respective numerical values for water [2 Marks]
- i) Latent Heat of Fusion [2 Marks]
- ii) Latent Heat of Vaporization
- b) A calorimeter contains 0.30 kg of water at 12°C . When poured in, the temperature of the mixture is found to be 52°C . What is the heat capacity of the calorimeter? [5 Marks]
- c) A piece of iron of mass 0.20 kg is heated to 64°C and then dropped gently into 0.15 kg of water at 16°C . If the temperature of the mixture is 22°C , what is the specific heat capacity of iron? [5 Marks]

- d) How much heat is given out when 50g of steam at 100°C cool to water at 28°C? (Specific latent heat of vapourization of water = $2.3 \times 10^6 \text{ J/kg}$). [6 Marks]

QUESTION FOUR [20 Marks]

- a) What is Absolute Temperature?[2 Marks]
- b) Find the pressure on a diver working at the depth of 10m in the sea on a day when the barometer stands at 750mmHg. (Density of sea water = 1050 kgm^{-3}). Express your answer in (a) mmHg and (b) Nm^{-2} [6 Marks]
- c) A balloon contains 1.5 m^3 of helium at a pressure of 100 kPa and at a temperature of 27 °C. If the pressure is increased to 250 kPa at a temperature of 127 °C, calculate the new volume of the balloon. [3 Marks]
- d) Explain the three gas laws in relation to kinetic theory of gases [9 Marks]

QUESTION FIVE [20 Marks]

- a) Define (i) tensile stress (ii) tensile strain (iii) young modulus. What are the units and dimensions of each of the above? [3 Marks]
- b) A 4.0 m long copper wire of cross-sectional area 1.2 cm^2 is stretched by a force of $4.8 \times 10^3 \text{ N}$. If Young modulus is $Y = 1.2 \times 10^{11} \text{ N/M}^2$, calculate (i) stress (ii) strain (iii) increase in length of the wire.[6 Marks]
- c) i) The surface tension of soap solution is $2.0 \times 10^{-2} \text{ N/m}$. How much work will be done in making a bubble of diameter 2.0 cm by blowing?[3 Marks]
- iii) A big drop is formed by coalescing 1000 small droplets of water. What will be the change in surface energy?[4 Marks]
- d) A metal plate of area $2.5 \times 10^{-4} \text{ m}^2$ is placed on a $0.25 \times 10^{-3} \text{ m}$ thick layer of castor oil. If a force of 2.5 N is needed to move the plate with a velocity $3 \times 10^{-2} \text{ m s}^{-1}$, calculate the coefficient of viscosity of castor oil. [4 Marks]

.....**END**.....