



# KIBABII UNIVERSITY

UNIVERSITY EXAMINATIONS  
2019/2020 ACADEMIC YEAR

SECOND YEAR SECOND SEMESTER  
MAIN EXAMINATIONS

FOR THE DEGREE OF  
B.SC RENEWABLE ENERGY AND BIOFUELS TECHNOLOGY

COURSE CODE: REN 221

COURSE TITLE: THERMODYNAMICS I

DURATION: 2 HOURS

DATE: 4/2/2021

TIME: 2:00-4:00PM

---

---

**INSTRUCTIONS TO CANDIDATES**

- (i) Answer **Question 1 (Compulsory)** and any other **TWO** questions
- (ii) All symbols have their usual meaning
- (iii) Use steam tables provided

This paper consists of **3** printed pages. Please Turn Over



KIBU observes ZERO tolerance to examination cheating

**QUESTION ONE (Compulsory) – 30 Marks**

- a) State the Non-Flow Energy Equation. (4 Marks)
- b) Steam at 110 bar has a specific volume of  $0.0196 \text{ m}^3/\text{kg}$ . Find the following properties:
- i) Temperature. (3 Marks)
  - ii) Internal energy. (3 Marks)
- c) Given steam at 0.5MPa with an enthalpy of 2.4MJ/kg, determine the:
- (i) Dryness fraction. (5 Marks)
  - (ii) Specific volume. (2 Marks)
  - (iii) Internal energy. (2 Marks)
- d) Show that for a perfect gas the following specific heats can be expressed as shown below:
- (i)  $C_v = \frac{R}{\gamma-1}$  (4 Marks)
  - (ii)  $C_p = \frac{\gamma R}{\gamma-1}$  (3 Marks)
- (e) Give two conditions for a thermodynamic equilibrium. (4 Marks)

**QUESTION TWO (20 Marks)**

Steam at a pressure of  $1.28 \text{ MN/m}^2$  enters a turbine with a velocity of  $26 \text{ m/s}$  and a specific volume of  $0.14 \text{ m}^3/\text{kg}$ . After a steady flow through the turbine the steam leaves at a pressure of  $38 \text{ kN/m}^2$ , a velocity of  $80 \text{ m/s}$  and a specific volume of  $4 \text{ m}^3/\text{kg}$ . The internal energy of steam leaving the turbine is  $260 \text{ kJ/kg}$  less than that of the steam entering the turbine. Heat is lost to the surroundings at a rate of  $0.3 \text{ kJ/s}$ . The steam flow is  $0.4 \text{ kg/s}$ . Calculate the:

- a) Power developed by the turbine. (14 Marks)
- b) Inlet and outlet cross-sectional areas. (6 Marks)

### QUESTION THREE (20 Marks)

A fluid at  $6 \times 10^4 \text{N/m}^2$ , occupying a volume of  $0.3 \text{m}^3$  is compressed reversibly to a pressure of  $0.24 \times 10^6 \text{N/m}^2$  according to a law  $pv^n = \text{constant}$ . The fluid is then heated reversibly at a constant volume until the pressure is  $38 \times 10^4 \text{N/m}^2$  and a specific volume of  $0.3 \text{m}^3/\text{kg}$ . The fluid is expanded reversibly according to the law  $pv^2 = \text{constant}$  to the initial state.

- Show the process on a p-v diagram. **(3 Marks)**
- Calculate the mass of the fluid. **(5 Marks)**
- Determine the value of n in the first process. **(5 Marks)**
- Calculate the net work done on or by fluid in the cycle. **(7 Marks)**

### QUESTION FOUR (20 Marks)

A fluid whose mass is 1kg is contained in a cylinder. The fluid is expanded reversibly behind a piston according to a law  $pv^2 = \text{constant}$  until its initial volume is doubled. The fluid is then cooled reversibly under constant pressure until the piston is at its original position; heat is then supplied reversibly with the piston firmly locked in position until the pressure rises to its initial value.

- Show the process on a P-V diagram. **(4 Marks)**
- If the initial pressure and volume are 18bar and  $0.04 \text{m}^3$  respectively, calculate the net work done on or by the fluid. **(16 Marks)**

### QUESTION FIVE (20 Marks)

- Show from first principles that the heat flow in a polytropic process is given by:

$$Q = \left( \frac{\gamma - n}{\gamma - 1} \right) W$$

**(10 Marks)**

- Carbon dioxide at 1 bar is compressed reversibly until the pressure is 6 bar according to the law  $PV^{1.4} = \text{constant}$ . If the initial specific volume is  $0.6 \text{m}^3/\text{kg}$ , calculate the:
  - Final temperature. **(3 Marks)**
  - Work done on the gas. **(2 Marks)**
  - Heat flow to or from the cylinder walls. **(5 Marks)**