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**KIBABII UNIVERSITY**  
**UNIVERSITY EXAMINATIONS**  
**2017/2018 ACADEMIC YEAR**  
**THIRD YEAR FIRST SEMESTER**  
**SUPPLIMENTARY / SPECIAL EXAMINATION**

**FOR THE DEGREE OF B.S.C**

**COURSE CODE:** SCH 350

**COURSE TITLE:** ENVIRONMENTAL CHEMISTRY

**DATE:** 10/10/2018

**TIME:** 11:30 – 1:30PM

**INSTRUCTION TO CANDIDATES**

Answer all questions

Observe zero tolerance to examination cheating

### Question one

- a. State the most important properties of pollutants that help in predicting the environmental behaviour of the pollutant (3mks)
- b. What are the important properties/ parameters of water that influences pollutants behaviour among different locations (5mks)
- c. State the important properties of soil that influence pollutants behaviour among different locations (4mks)
- d. Explain the following commonly used terms in environmental chemistry (10mks)
- i. Receptor
  - ii. Sink
  - iii. Dissolved oxygen (DO)
  - iv. Chemical oxygen demand (COD)
  - v. Biological oxygen demand (BOD)

### Question two

- a. The environment consists of various segments
- i. Name the four segments of the environment (2mks)
  - ii. Explain the vital role played by the atmosphere in the survival of life in this planet (6mks)
- c. Complete the following table that shows the major regions of atmosphere and their characteristics.

Fill in the blank space A, B, C, D and E

(5mks)

Region	Altitude range km	Temperature range °C	Significant chemical species
Troposphere	<b>A</b>	15 to -56	N <sub>2</sub> , O <sub>2</sub> , CO <sub>2</sub> , H <sub>2</sub> O
<b>B</b>	11-50	-56 to -2	<b>C</b>
<b>D</b>	50-85	-2 to -92	O <sub>2</sub> <sup>+</sup> , NO <sup>+</sup>
Thermosphere	<b>E</b>	-92 to 1200	O <sub>2</sub> <sup>+</sup> , O <sup>+</sup> , NO <sup>+</sup>

### ***Question three***

- a. As an environmental chemist. Give three reasons Why you think (1, 1, 1-trichloro-2, 2-di (4-phenyl ethane) DDT was banned for use. (6mks)
- b. State and briefly describe any three important environmental chemical reactions that can remove pollutants from aqueous environment (6mks)

### ***Question four***

- a. Discuss the role of ozone in the atmosphere 4mks
- b. 90% of ozone layer in the atmosphere can be found in the stratosphere. Discuss the formation of Actical and Antactical holes in the North Pole and South Pole in the stratosphere (6mks)
- c. what is the benefits of climate change (2mks)
- d. Define the green house effect (3mks)
- e. Answer the following questions
- i. What is ozone hole (3mks)
- ii. What causes ozone hole (3mks)
- iii. What does ozone hole have to do with climate change (2mks) .